Acta Crystallographica Section E **Structure Reports** Online

ISSN 1600-5368

# Alan L. Raine, Craig M. Williams and Paul V. Bernhardt\*

Department of Chemistry, The University of Queensland, Brisbane, Queensland 4072, Australia

Correspondence e-mail: p.bernhardt@mailbox.uq.edu.au

### Key indicators

Single-crystal X-ray study T = 296 KMean  $\sigma(C-C) = 0.006 \text{ Å}$ R factor = 0.057 wR factor = 0.194 Data-to-parameter ratio = 14.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

The title compound,  $C_{18}H_{20}$ , provides the first structurally characterized example of an alkynyl-substituted adamantane free from cocrystallized solvent or guest molecules.

1-(Phenylethynyl)adamantane

Received 28 August 2002 Accepted 21 November 2002 Online 30 November 2002

## Comment

Due to the increasing number of nano-machines (Balzani et al., 2000; Kelly, 2001) and nano-devices (Rukavishnikov et al., 1999) containing acetylene units attached to bulky caged carbocycles, we have been investigating new routes to achieving one-step synthesis of such systems. Recently, we discovered that the title compound, (I), could be obtained, via a novel metal-mediated process, from phenylacetylene and 1iodoadamantane (Williams & Raine, 2002). It comprises adamantyl and phenyl groups bridged by a single acetylene residue.



Although structurally characterized examples of substituted adamantanes abound (more than 150 examples in the Cambridge Structural Database; Allen, 2002), there is but one molecule where the substituent is an alkyne attached to the bridgehead C atom. This compound, namely the symmetrical dumb-bell-shaped molecule 1,8-bis(1-adamantyl)-1,3,5,7octatetrayne, (II), has been found to exhibit a number of interesting structural and non-linear optical properties. Significantly, all structures of (II) have been found to include a cocrystallized guest molecule, such as 2-butanone, crocetin dialdehyde, cyclohexanol or *trans*- $\beta$ -8'-apocarotenal (Müller et al., 2000). It was found that (II) exhibits a distinctly bowed conformation of its nominally linear tetra-yne moiety, and the cocrystallized guest molecule appears to have an important influence on the observed conformation of its host (II). We were interested in the conformation of the simple analogue (I), which bears the same ethynyladamantyl group and, to this end, we have been successful in crystallizing (I) in a solventfree form.

Compound (I) comprises adamantyl and phenyl groups bridged by a single acetylene residue. There is little apparent strain in the molecule and the bond lengths (Table 1) are as expected for a compound of its type. The adamantyl (1.50-1.54 Å) and phenyl C–C bond lengths (1.36-1.40 Å) are typical for these groups. The (adamantyl)C1-C11-C12 and

Printed in Great Britain - all rights reserved

© 2002 International Union of Crystallography



ORTEP-3 plot (Farrugia, 1997) of (I), with ellipsoids at the 30% probability level.

C11-C12-C13(phenyl) angles (Table 1) are both close to linear, although the latter exhibits a small deviation therefrom. There are no hydrogen-bonding interactions in hydrocarbon (I). The closest intermolecular contacts are H9B···H18(-x + 1,  $y - \frac{1}{2}$ ,  $-z + \frac{1}{2}$ ) (2.44 Å), H2A···H14(x,  $-y + \frac{1}{2}$ ,  $z + \frac{1}{2}$ ) (2.53 Å) and H6B···H7A(-x + 1, -y, -z) (2.53 Å).

In conclusion, in the absence of any cocrystallized guest molecule, we have found that the nominally rod-shaped (I) exhibits an essentially undistorted conformation.

## **Experimental**

The synthesis of the title compound will be reported separately (Williams & Raine, 2002). Crystals were obtained by slow evaporation of a petroleum spirit (313–333 K) solution of the compound.

Crystal data

```
\begin{array}{l} C_{18}H_{20} \\ M_r = 236.34 \\ \text{Monoclinic, } P_{21}/c \\ a = 9.050 \ (1) \text{ Å} \\ b = 14.224 \ (1) \text{ Å} \\ c = 11.081 \ (1) \text{ Å} \\ \beta = 107.986 \ (9)^{\circ} \\ V = 1356.7 \ (2) \text{ Å}^3 \\ Z = 4 \end{array}
```

## Data collection

Enraf-Nonius CAD-4 diffractometer  $\omega$ -2 $\theta$  scans Absorption correction:  $\psi$  scan (North *et al.*, 1968)  $T_{min} = 0.911$ ,  $T_{max} = 0.991$ 2542 measured reflections 2385 independent reflections 1381 reflections with  $I > 2\sigma(I)$ 

### Refinement

Refinement on $F^2$
$R[F^2 > 2\sigma(F^2)] = 0.057$
$wR(F^2) = 0.194$
S = 1.15
2385 reflections
164 parameters
H-atom parameters constrained

 $D_x = 1.157 \text{ Mg m}^{-3}$ Mo K\alpha radiation Cell parameters from 25 reflections  $\theta = 9.8-13.9^{\circ}$  $\mu = 0.07 \text{ mm}^{-1}$ T = 296 (2) K Prism, colourless  $0.50 \times 0.27 \times 0.10 \text{ mm}$  $R_{int} = 0.028$ 

$\theta_{\rm max} = 25.0^{\circ}$
$h = 0 \rightarrow 10$
$k = 0 \rightarrow 16$
$l = -13 \rightarrow 12$
3 standard reflections
frequency: 120 min
intensity decay: 4%

 $w = 1/[\sigma^{2}(F_{o}^{2}) + (0.038P)^{2} + 2.0561P]$ where  $P = (F_{o}^{2} + 2F_{c}^{2})/3$   $(\Delta/\sigma)_{max} < 0.001$   $\Delta\rho_{max} = 0.17 \text{ e } \text{\AA}^{-3}$   $\Delta\rho_{min} = -0.17 \text{ e } \text{\AA}^{-3}$ Extinction correction: SHELXL Extinction coefficient: 0.023 (3)

Table	1	

Selected	geometric	parameters	(Å,	°).
----------	-----------	------------	-----	-----

C1-C11	1.477 (5)	C8-C9	1.523 (6)
C1-C6	1.534 (5)	C8-C10	1.524 (6)
C1-C7	1.541 (5)	C11-C12	1.190 (5)
C1-C2	1.542 (5)	C12-C13	1.433 (5)
C2-C3	1.523 (5)	C13-C18	1.393 (5)
C3-C10	1.506 (6)	C13-C14	1.398 (5)
C3-C4	1.517 (6)	C14-C15	1.379 (5)
C4-C5	1.524 (6)	C15-C16	1.369 (6)
C5-C9	1.521 (6)	C16-C17	1.376 (6)
C5-C6	1.530 (6)	C17-C18	1.380 (5)
C7-C8	1.521 (6)		
C11-C1-C6	110.0 (3)	C8-C7-C1	110.2 (3)
C11-C1-C7	109.8 (3)	C7-C8-C9	109.5 (4)
C6-C1-C7	109.0 (3)	C7-C8-C10	109.8 (4)
C11-C1-C2	111.5 (3)	C9-C8-C10	109.2 (4)
C6-C1-C2	108.4 (3)	C5-C9-C8	109.4 (4)
C7-C1-C2	108.1 (3)	C3-C10-C8	109.4 (3)
C3-C2-C1	109.4 (3)	C12-C11-C1	179.0 (4)
C10-C3-C4	110.0 (4)	C11-C12-C13	177.8 (5)
C10-C3-C2	110.1 (3)	C18-C13-C14	117.8 (3)
C4-C3-C2	110.3 (3)	C18-C13-C12	121.5 (3)
C3-C4-C5	108.8 (3)	C14-C13-C12	120.7 (4)
C9-C5-C4	110.0 (4)	C15-C14-C13	120.9 (4)
C9-C5-C6	109.9 (4)	C15-C16-C17	119.9 (4)
C4-C5-C6	108.8 (4)	C16-C17-C18	120.3 (4)
C5-C6-C1	110.1 (3)	C17-C18-C13	120.7 (4)

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994); cell refinement: *CAD-4 EXPRESS*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS8*6 (Sheldrick, 1985); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

### References

- Allen, F. H. (2002). Acta Cryst. B58, 380-388.
- Balzani, V., Credi, A., Raymo, F. M. & Stoddart, J. F. (2000). *Angew. Chem. Int. Ed.* **39**, 3348–3391.
- Enraf-Nonius (1994). CAD-4 EXPRESS. Enraf-Nonius, Delft, The Netherlands.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
- Harms, K. & Wocadlo, S. (1995). XCAD4. University of Marburg, Germany. Kelly, T. R. (2001), Acc. Chem. Res. 34, 514–522.
- Müller, T., Hulliger, J., Seichter, W., Weber, E., Weber, T. & Wübbenhorst, M. (2000). *Chem. Eur. J.* **6**, 54–61.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). Acta Cryst. A24, 351–359.
- Rukavishnikov, A. V., Phadke, A., Lee, M. D., LaMunyon, D. H., Petukhov, P. A. & Keana, J. F. W. (1999). *Tetrahedron Lett.* **40**, 6353–6356.
- Sheldrick, G. M. (1985). SHELXS86. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXL97. Release 97-2. University of Göttingen, Germany.
- Williams, C. M. & Raine, A. L. (2002). In preparation.